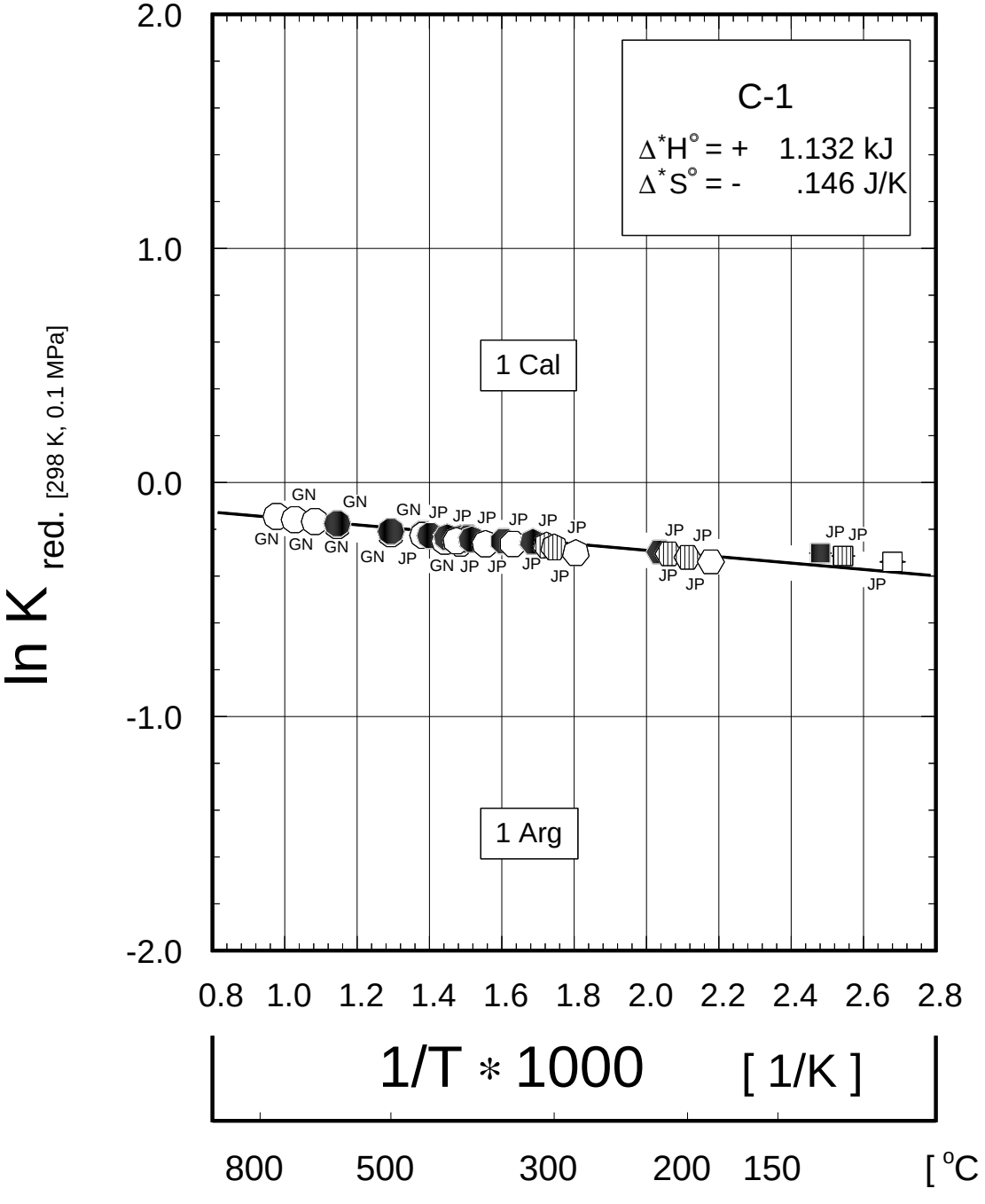
C-1
Δ*H° = + 1.132 kJ
Δ*S° = - .146 J/K
1 Cal
1 Arg
ln K red. [298 K, 0.1 MPa]
2.0
1.0
0.0
-1.0
-2.0
0.8 1.0 1.2 1.4 1.6 1.8 2.0 2.2 2.4 2.6 2.8
1/T * 1000 [1/K]
800 500 300 200 150 [°C]
GN
JP